

OPCW

Organisation for the Prohibition of Chemical Weapons

Illuminating Chemical Reactivity

an event guaranteed to brighten up your day...

Science for Diplomats at EC-91 The Hague, 9 July 2019

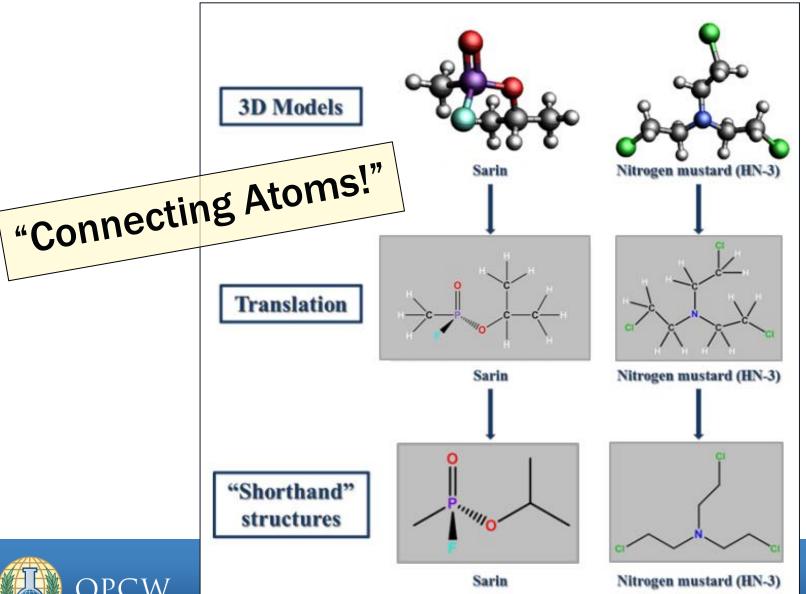
Starring

Dr Marc-Michael Blum, Head OPCW Laboratory

With supporting cast

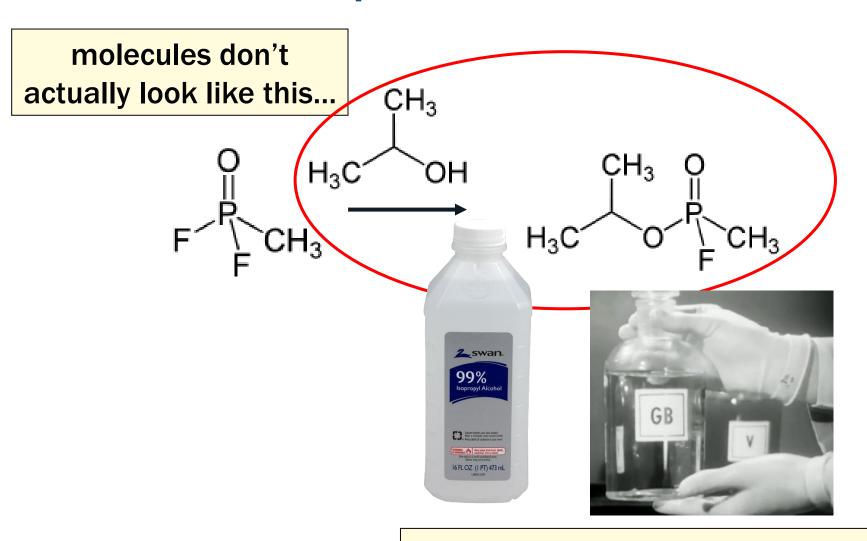
Mr Cheng Tang (SAB Chair), Mr Lucas Benderitter (OSP), Mr Peter Brud (OSP), Dr Jonathan E. Forman (Science Policy Adviser and SAB Secretary), Ms Giovanna Pontes (OSP), Ms Ayah Wafi (OSP) and special guest Ms Andrea Dymytrova

Chemistry Lessons





The Elephant in the Room...

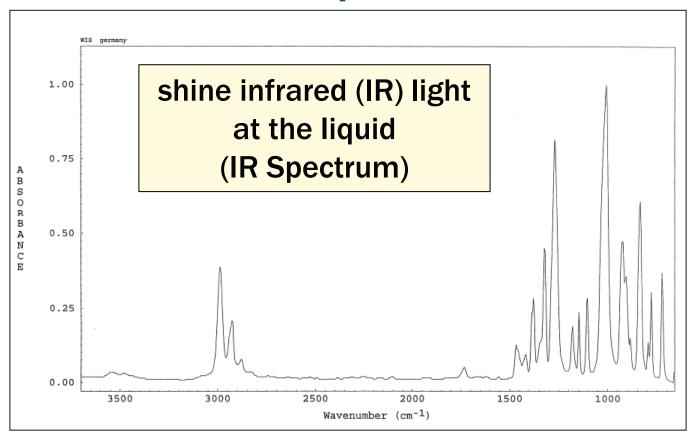




colourless liquids...

How do you know which one is which?

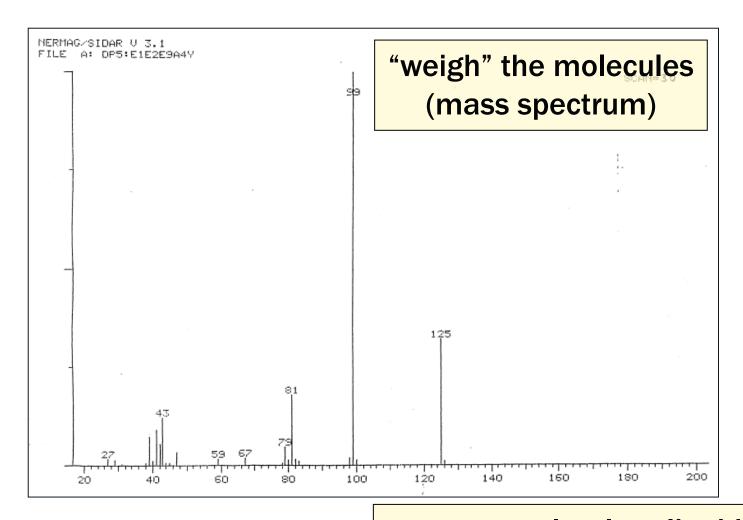
The Elephant in the Room...



colourless liquids...
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The Elephant in the Room...

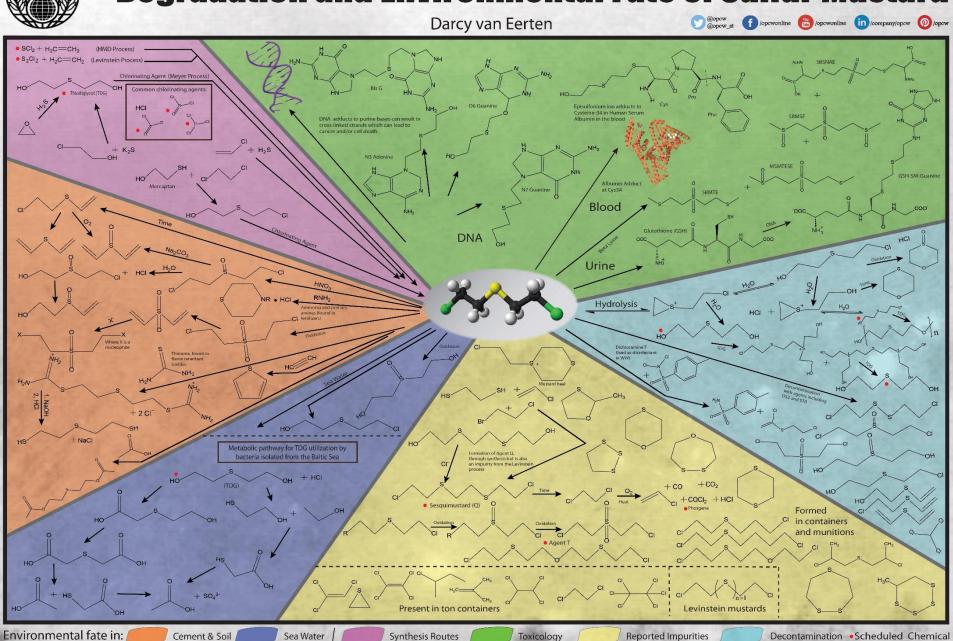


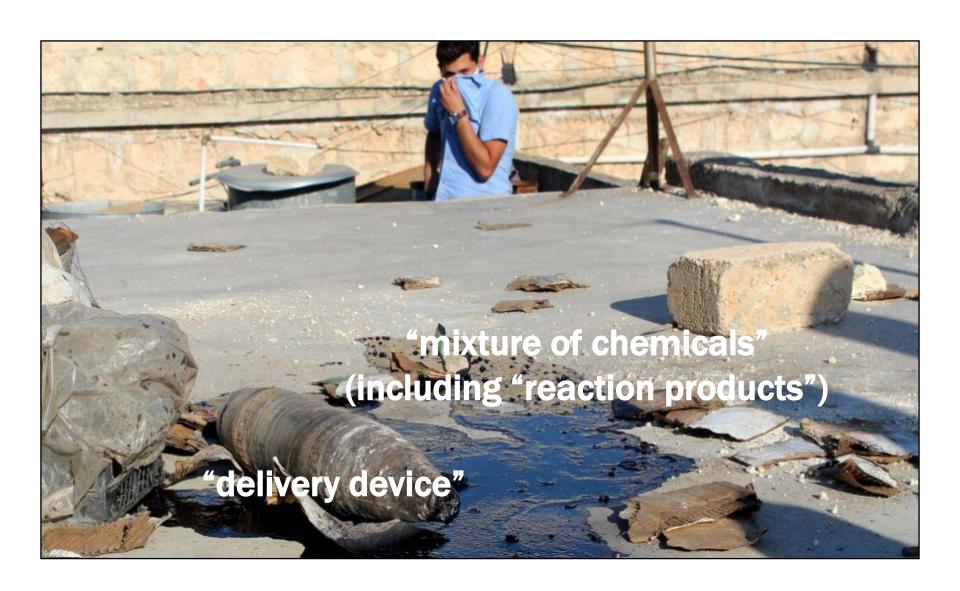


colourless liquids...

How do you know which one is which?

Degradation and Environmental Fate of Sulfur Mustard







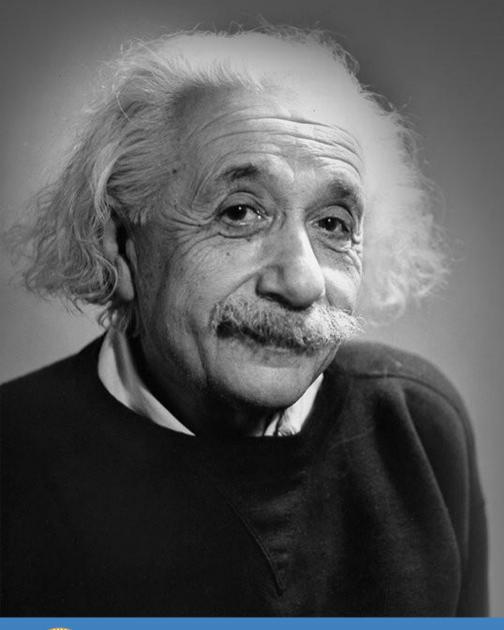
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Organisation for the Prohibition of Chemical Weapons

Chemical reactivity

or why chemistry is basically the same as politics

Marc-Michael Blum, Ph.D. Head, OPCW Laboratory



Politics is more difficult than physics

Albert Einstein







Chemistry:

Chemistry is the study of matter, its properties, how and why substances combine or separate to form other substances, and how substances interact with energy.



The Chemical Weapons Convention is quite focused on chemicals themselves:

- Declarations based on production, consumption and/or transfers of chemicals
- Annex on Chemicals of the CWC listing those chemicals for which special verification measures are in place
- Sampling & Analysis is conducted to confirm the presence or absence of a CWC relevant chemical
- •

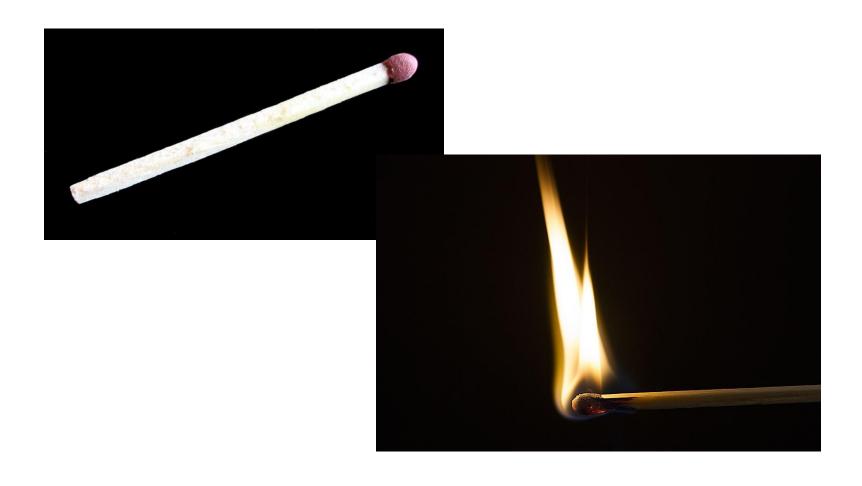


But we also deal with REACTIONS of chemicals:

- We discuss the meaning of "production by synthesis"
- We discuss the the productions of DOCs via biomediated processes
- Sampling & Analysis is looking for precursors and degradation products of chemical agents in IAUs
- Reaction products of agents with biomolecules ("adducts") are valuable biomarkers and important in biomedical verification
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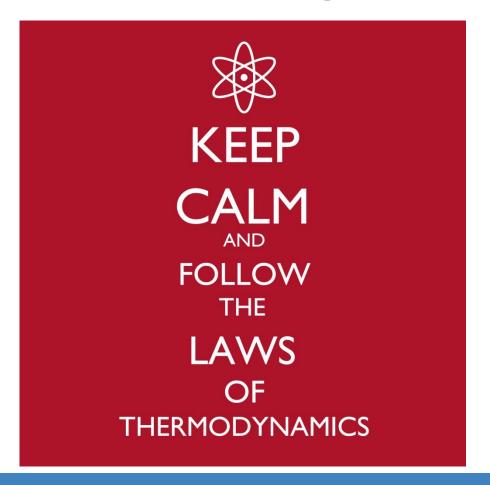


So why do chemicals react with each other?





What are the laws that determine in what direction a chemical reaction is proceeding?





The <u>first law</u> of thermodynamics is a version of the law of conservation of energy, adapted for thermodynamic systems. The law of conservation of energy states that the total energy of an isolated system is constant; energy can be transformed from one form to another, but can be neither created nor destroyed.



The <u>second law</u> of thermodynamics states that the total entropy of an isolated system can never decrease over time. The total entropy of a system and its surroundings can remain constant in ideal cases where the system is in thermodynamic equilibrium. In all processes that occur, including spontaneous processes, the total entropy of the system and its surroundings increases and the process is irreversible in the thermodynamic sense. The increase in entropy accounts for the irreversibility of natural processes, and the asymmetry between future and past.



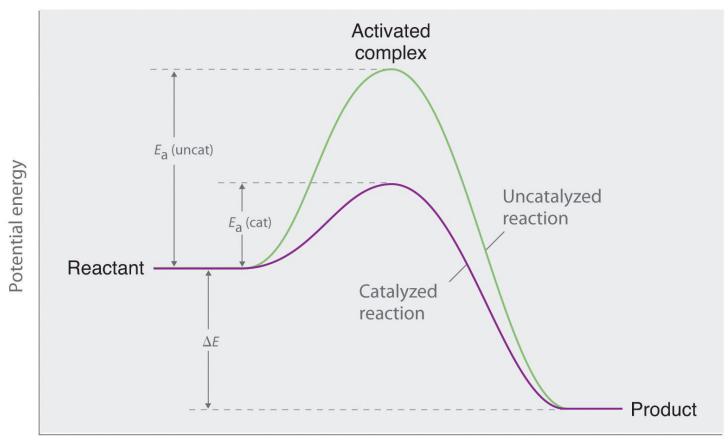
Chemical equilibrium

Thermodynamics determine if A and B or C and D are the favoured products and where the equilibrium is located. It does NOT determine reaction rates.

$$A+B \xrightarrow{\longrightarrow} C+D$$
 $A+B \xrightarrow{\longrightarrow} C+D$



Activation Energy and reaction rates



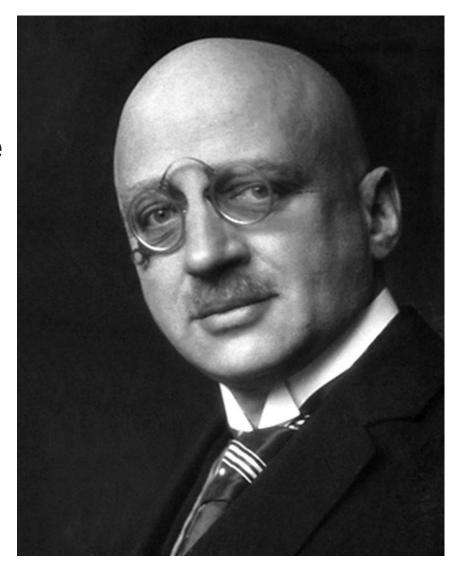
Reaction coordinate



Fritz Haber (1868-1934)

"Father" of chemical warfare in World War I

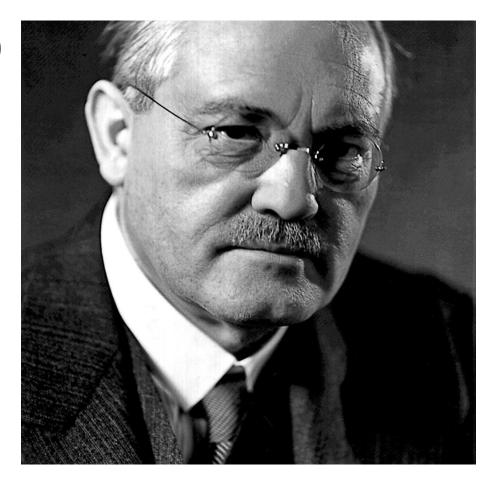
But also Nobel Laureate in Chemistry 1918



Carl Bosch (1874-1940)

Chemist and Industrialist

Nobel Laureate in Chemistry 1931





The Haber-Bosch process

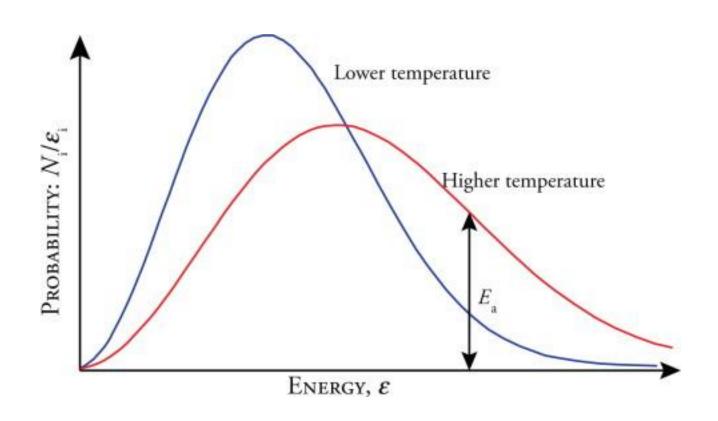
Making ammonia with nitrogen from the air

$$N_2 + 3 H_2 \rightleftharpoons 2 NH_3$$

- Nitrogen is a very stable and unreactive molecule.
- Process has high activation barrier.
- Raising the temperature makes reaction faster but shifts equilibrium from ammonia towards starting products



Reaction rate and temperature





The Haber-Bosch process

The solution:

- Lowering activation energy a using metal catalyst
- Temperature for reasonable reaction rates now lower but still favouring the starting products
- Shifting the equilibrium towards ammonia by applying high pressure.



The Haber-Bosch process

Major technological breakthrough (high pressure reactions in industry). Uses about 1-2% of world energy consumption.







- Enabled Imperial Germany to continue to fight World War I despite being blocked from accessing natural nitrate deposits Enabled the mass production of nitrogen fertilizers
- Enabling massive growth of agricultural production
- Without the Haber process the current world population would not be possible
- About half of all nitrogen atoms in the human body are derived from air nitrogen via the Haber process.



Chemical warfare and reactivity

 Nerve agents should have high reactivity with the biological target (Acetylcholinesterase) but low reactivity towards water (hydrolytic stability). Fluoridates better than chloridates.

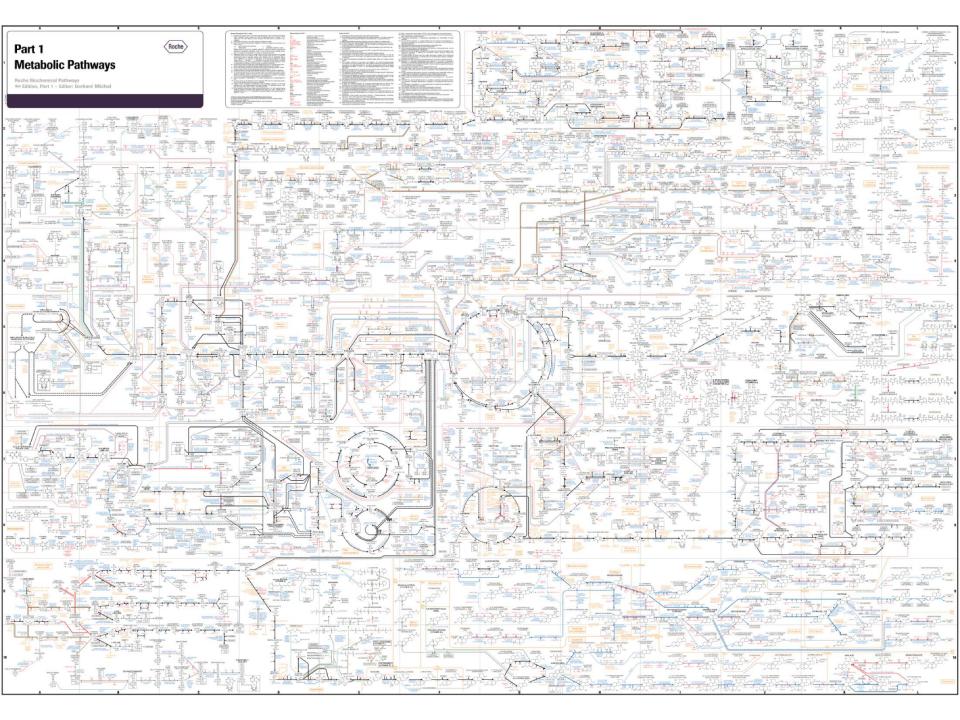


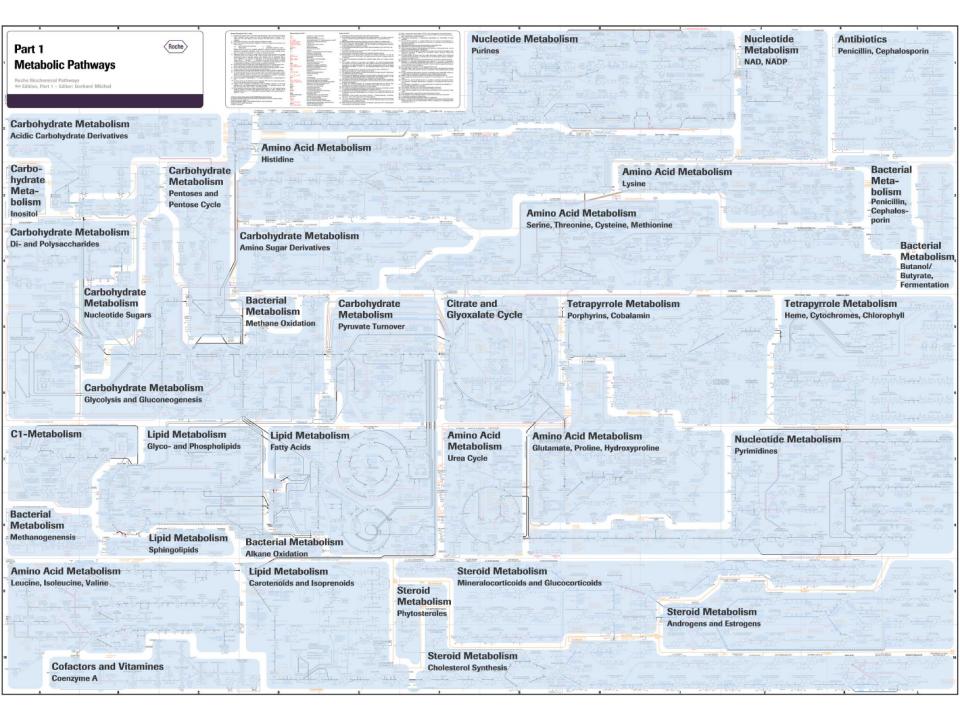
But back to Chemistry and Politics....

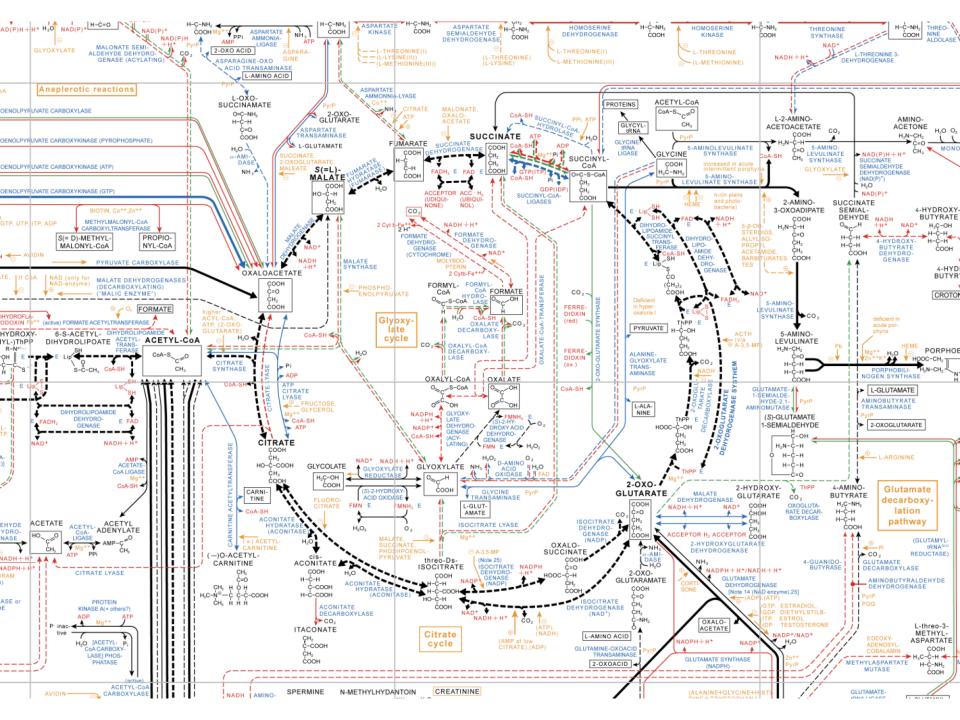
How can this be similar to politics? Clearly defined start and end states? Predictable reaction rates and equilibrium? No room for negotiations?

Chemistry can be a little bit more complicated.....











OPCW

منظمة حظر الأسلحة الكيميائية

禁止化学武器组织

Organisation for the Prohibition of Chemical Weapons

Organisation pour l'Interdiction des Armes Chimiques

Организация по запрещению химического оружия

Organización para la Prohibición de las Armas Químicas